## **EAST Search History**

Ref #	Hits	Search Query	DBs	Default Operator	Plurals	Time Stamp
L1	854	549/531 or 203/23 or 203/86	US-PGPUB; USPAT; USOCR; FPRS; EPO; JPO; DERWENT; IBM_TDB	OR	ON	2007/09/20 15:55
L2	1	l1 and alkene and hydroperoxide and "alkene oxide" and solvent and compressed	US-PGPUB; USPAT; USOCR; FPRS; EPO; JPO; DERWENT; IBM_TDB	OR	ON	2007/09/20 16:11
L3	236	l1 and (propylene or propene) and (hydroperoxide or "hydrogen peroxide")and "propylene oxide"	US-PGPUB; USPAT; USOCR; FPRS; EPO; JPO; DERWENT; IBM_TDB	OR	ON	2007/09/20 16:11
L4	19	I3 and (compressed or compression)	US-PGPUB; USPAT; USOCR; FPRS; EPO; JPO; DERWENT; IBM_TDB	OR	ON	2007/09/20 16:12
L5	10	alkene and hydroperoxide and "alkene oxide" and solvent and compressed	US-PGPUB; USPAT; USOCR; FPRS; EPO; JPO; DERWENT; IBM_TDB	OR	ON	2007/09/20 16:11
L6	14809	(propylene or propene) and (hydroperoxide or "hydrogen peroxide")and "propylene oxide"	US-PGPUB; USPAT; USOCR; FPRS; EPO; JPO; DERWENT; IBM_TDB	OR	ON	2007/09/20 16:11
L7	2975	l6 and (compressed or compression)	US-PGPUB; USPAT; USOCR; FPRS; EPO; JPO; DERWENT; IBM_TDB	OR	ON	2007/09/20 16:13

## **EAST Search History**

L8	1084	I7 and (distill or distillation)	US-PGPUB; USPAT; USOCR; FPRS; EPO; JPO; DERWENT; IBM_TDB	OR	ON	2007/09/20 16:14
L9	11	I8 and "heat exchange"	US-PGPUB; USPAT; USOCR; FPRS; EPO; JPO; DERWENT; IBM_TDB	OR	ON	2007/09/20 16:51
L10	41	I8 and (decompression or refrigeration)	US-PGPUB; USPAT; USOCR; FPRS; EPO; JPO; DERWENT; IBM_TDB	OR	ON	2007/09/20 17:43
L11	15	"5744619"	US-PGPUB; USPAT; USOCR; FPRS; EPO; JPO; DERWENT; IBM_TDB	OR	ON	2007/09/20 17:43
S1	10	alkene and hydroperoxide and "alkene oxide" and solvent and compressed	US-PGPUB; USPAT; USOCR; FPRS; EPO; JPO; DERWENT; IBM_TDB	OR	ON	2007/09/20 15:56
53	1	10/553516	US-PGPUB; USPAT; USOCR; FPRS; EPO; JPO; DERWENT; IBM_TDB	OR	ON	2007/09/20 14:51

Welcome to STN International! Enter x:x
LOGINID:SSPTADEG1625

PASSWORD:

TERMINAL (ENTER 1, 2, 3, OR ?):2

```
NEWS
                 Web Page for STN Seminar Schedule - N. America
NEWS
         JUL 02
                 LMEDLINE coverage updated
NEWS
                 SCISEARCH enhanced with complete author names
         JUL 02
NEWS
         JUL 02
                 CHEMCATS accession numbers revised
NEWS
         JUL 02
                 CA/CAplus enhanced with utility model patents from China
         JUL 16
                 CAplus enhanced with French and German abstracts
NEWS
                 CA/CAplus patent coverage enhanced
NEWS
         JUL 18
NEWS
         JUL 26
                 USPATFULL/USPAT2 enhanced with IPC reclassification
NEWS 9
         JUL 30
                 USGENE now available on STN
                 CAS REGISTRY enhanced with new experimental property tags
NEWS 10
         AUG 06
NEWS 11
         AUG 06
                 BEILSTEIN updated with new compounds
                 FSTA enhanced with new thesaurus edition
NEWS 12
         AUG 06
                 CA/CAplus enhanced with additional kind codes for granted
NEWS 13
         AUG 13
                 patents
         AUG 20
                 CA/CAplus enhanced with CAS indexing in pre-1907 records
NEWS 14
NEWS 15
         AUG 27
                 Full-text patent databases enhanced with predefined
                 patent family display formats from INPADOCDB
NEWS 16
         AUG 27
                 USPATOLD now available on STN
NEWS 17
         AUG 28
                 CAS REGISTRY enhanced with additional experimental
                 spectral property data
                 STN AnaVist, Version 2.0, now available with Derwent
NEWS 18
         SEP 07
                 World Patents Index
NEWS 19
         SEP 13
                 FORIS renamed to SOFIS
NEWS 20
         SEP 13
                 INPADOCDB enhanced with monthly SDI frequency
         SEP 17
NEWS 21
                 CA/CAplus enhanced with printed CA page images from
                 1967-1998
NEWS 22
         SEP 17
                 CAplus coverage extended to include traditional medicine
                 patents
```

NEWS EXPRESS 19 SEPTEMBER 2007: CURRENT WINDOWS VERSION IS V8.2, CURRENT MACINTOSH VERSION IS V6.0c(ENG) AND V6.0Jc(JP), AND CURRENT DISCOVER FILE IS DATED 19 SEPTEMBER 2007.

NEWS HOURS STN Operating Hours Plus Help Desk Availability
NEWS LOGIN Welcome Banner and News Items
NEWS IPC8 For general information regarding STN implementation of IPC 8

Enter NEWS followed by the item number or name to see news on that specific topic.

All use of STN is subject to the provisions of the STN Customer agreement. Please note that this agreement limits use to scientific research. Use for software development or design or implementation of commercial gateways or other similar uses is prohibited and may result in loss of user privileges and other penalties.

=> file caplus
COST IN U.S. DOLLARS

FULL ESTIMATED COST

SINCE FILE TOTAL ENTRY SESSION 0.21 0.21

FILE 'CAPLUS' ENTERED AT 17:27:37 ON 20 SEP 2007 USE IS SUBJECT TO THE TERMS OF YOUR STN CUSTOMER AGREEMENT. PLEASE SEE "HELP USAGETERMS" FOR DETAILS.
COPYRIGHT (C) 2007 AMERICAN CHEMICAL SOCIETY (ACS)

Copyright of the articles to which records in this database refer is held by the publishers listed in the PUBLISHER (PB) field (available for records published or updated in Chemical Abstracts after December 26, 1996), unless otherwise indicated in the original publications. The CA Lexicon is the copyrighted intellectual property of the American Chemical Society and is provided to assist you in searching databases on STN. Any dissemination, distribution, copying, or storing of this information, without the prior written consent of CAS, is strictly prohibited.

FILE COVERS 1907 - 20 Sep 2007 VOL 147 ISS 13 FILE LAST UPDATED: 19 Sep 2007 (20070919/ED)

Effective October 17, 2005, revised CAS Information Use Policies apply. They are available for your review at:

http://www.cas.org/infopolicy.html

=> s (alkene or propene or propylene) and epoxidation and ("liquid alkene" or "liquid propene" or "liquid propylene" or "condensed alkene" or "condensed propylene" or "condensed propene") and (hydroperoxide or "hydrogen peroxide")

36601 ALKENE

86730 ALKENES

99840 ALKENE

(ALKENE OR ALKENES)

75473 PROPENE

776 PROPENES

75809 PROPENE

(PROPENE OR PROPENES)

190676 PROPYLENE

305 PROPYLENES

190775 PROPYLENE

(PROPYLENE OR PROPYLENES)

14745 EPOXIDATION

247 EPOXIDATIONS

14780 EPOXIDATION

(EPOXIDATION OR EPOXIDATIONS)

26320 EPOXIDN

573 EPOXIDNS

26409 EPOXIDN

(EPOXIDN OR EPOXIDNS)

28147 EPOXIDATION

(EPOXIDATION OR EPOXIDN)

789750 "LIQUID"

137059 "LIQUIDS"

892387 "LIQUID"

("LIQUID" OR "LIQUIDS")

1093039 "LIQ"

103662 "LIOS"

1132706 "LIQ"

("LIQ" OR "LIQS")

1569236 "LIQUID"

("LIQUID" OR "LIQ")

```
36601 "ALKENE"
  86730 "ALKENES"
  99840 "ALKENE"
          ("ALKENE" OR "ALKENES")
     66 "LIQUID ALKENE"
          ("LIQUID"(W)"ALKENE")
789750 "LIQUID"
 137059 "LIQUIDS"
 892387 "LIQUID"
          ("LIQUID" OR "LIQUIDS")
1093039 "LIQ"
 103662 "LIQS"
1132706 "LIQ"
          ("LIQ" OR "LIQS")
1569236 "LIOUID"
          ("LIQUID" OR "LIQ")
  75473 "PROPENE"
    776 "PROPENES"
  75809 "PROPENE"
          ("PROPENE" OR "PROPENES")
     85 "LIQUID PROPENE"
          ("LIQUID" (W) "PROPENE")
 789750 "LIQUID"
 137059 "LIQUIDS"
 892387 "LIQUID"
          ("LIQUID" OR "LIQUIDS")
1093039 "LIO"
103662 "LIOS"
1132706 "LIO"
          ("LIQ" OR "LIQS")
1569236 "LIQUID"
          ("LIQUID" OR "LIQ")
 190676 "PROPYLENE"
    305 "PROPYLENES"
 190775 "PROPYLENE"
          ("PROPYLENE" OR "PROPYLENES")
    751 "LIQUID PROPYLENE"
          ("LIQUID" (W) "PROPYLENE")
 125182 "CONDENSED"
  36601 "ALKENE".
  86730 "ALKENES"
  99840 "ALKENE"
          ("ALKENE" OR "ALKENES")
      1 "CONDENSED ALKENE"
          ("CONDENSED" (W) "ALKENE")
 125182 "CONDENSED"
 190676 "PROPYLENE"
    305 "PROPYLENES"
 190775 "PROPYLENE"
          ("PROPYLENE" OR "PROPYLENES")
      5 "CONDENSED PROPYLENE"
          ("CONDENSED" (W) "PROPYLENE")
 125182 "CONDENSED"
  75473 "PROPENE"
    776 "PROPENES"
  75809 "PROPENE"
          ("PROPENE" OR "PROPENES")
      1 "CONDENSED PROPENE"
          ("CONDENSED"(W)"PROPENE")
  33681 HYDROPEROXIDE
  15336 HYDROPEROXIDES
  39996 HYDROPEROXIDE
          (HYDROPEROXIDE OR HYDROPEROXIDES)
1015605 "HYDROGEN"
   6041 "HYDROGENS"
```

1018965 "HYDROGEN" OR "HYDROGENS")
220699 "PEROXIDE"
48207 "PEROXIDES"
239693 "PEROXIDE"

("PEROXIDE" OR "PEROXIDES")

121842 "HYDROGEN PEROXIDE"

("HYDROGEN" (W) "PEROXIDE")

8 (ALKENE OR PROPENE OR PROPYLENE) AND EPOXIDATION AND ("LIQUID ALKENE" OR "LIQUID PROPENE" OR "LIQUID PROPYLENE" OR "CONDENSED ALKENE" OR "CONDENSED PROPYLENE" OR "CONDENSED PROPENE") AND (HYDROPEROXIDE OR "HYDROGEN PEROXIDE")

=> d l1 abs ibib

L1

L1 ANSWER 1 OF 8 CAPLUS COPYRIGHT 2007 ACS on STN

AB The method comprises epoxidizing liquid propylene with
liquid organic hydroperoxide in the presence of a catalyst, wherein
temperature of propylene gas introduced into the inlet of a compressor
to compress is higher than that saturation temperature The method
prevents the drain
formation with supplying the gas at the temperature which is higher than
dew-point temperature of the gas which is supplied to the compressor.

ACCESSION NUMBER: 2005/297624 CAPPUS
DOCUMENT NUMBER: 142:155703
TITLE: Method for production of propylene oxide
INVENTOR(S): Shinchara, Koji: Omae, Shunichi
SOURCE: Japan
SOURCE: Japan
DOCUMENT TYPE: Patent
LANGUAGE: Japanese
FAHILY ACC. NUM. COUNT: 1
PATENT INFORMATION:

PATENT NO.

JP 2005089404

PRIORITY APPLN. INFO.:

KIND ----APPLICATION NO.

JP 2003-327709
JP 2003-327709 DATE 20050407

DATE 20030919 20030919

=> d l1 2-8 abs ibib

```
ANSWER 2 OF 8 CAPLUS COPYRIGHT 2007 ACS on STN
A method is described for producing an epoxide (e.g., propylene oxide) comprising: (i) preparation of a stream (51) containing a
           ressed liquid alkene (e.g., propylene); (ii) expansion of at a least part of the stream (S1) by heat absorption and at least partial evaporation of the liquid alkene; (ii) reaction of the alkene obtained according to step (ii) with a. hydroperoxide (e.g., hydrogen peroxide) in the presence of at least one solvent (e.g., methanol) and at least one catalyst (e.g., titanium silicalite) to obtain a mixture containing the kide
and the solvent(s).
ACCESSION NUMBER:
DOCUMENT NUMBER:
                                                         2004:902364 CAPLUS
                                                         2004:902364 CAPLUS
141:380278
Method for producing an epoxide
Goebbel, Hans-Georg; Bassler, Peter: Teles, Joaquim
Henrique; Rudolf, Peter
BASF Aktiengesellschaft, Germany
PCT Int. Appl., 27 pp.
CODEN: PIXXD2
Patent
German
INVENTOR(S):
PATENT ASSIGNEE(S):
SOURCE:
DOCUMENT TYPE:
LANGUAGE:
LANGUAGE:
FAMILY ACC. NUM. COUNT:
PATENT INFORMATION:
        PATENT NO.
                                                          KIND
                                                                         DATE
                                                                                                     APPLICATION NO.
                                                                                                                                                           DATE
IN 2005CN02639
PRIORITY APPLN. INFO.:
                                                                                                     WO 2004-EP4077
                                                                        THERE ARE 5 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE
REFERENCE COUNT:
FORMAT
```

```
ANSWER 4 OF 8 CAPLUS COPYRIGHT 2007 ACS on STN
The invention relates to a method of regenerating a solid catalyst used
for an epoxidn. of propylene and an organic peroxide such
as cumene hydroperoxide in a reactor filled with the solid
catalyst, wherein a liquid such as propylene passes through the
reactor at a temperature higher than the maximum temperature of the
epoxidn. by
25° to regenerate the solid catalyst.
ACCESSION NUMBER: 2002:704699 CAPLUS
DOCUMENT NUMBER: 137:222566
Method of regenerating solid catalyst
ITLE: Method of regenerating solid catalyst
ITMENTOR(S): Tsuji, Junpsi, Osaki, Shunichi
Source: Source: Jon. Kokai Tokkyo Koho, 3 pp.
CODEN: JKXXAF
DOCUMENT TYPE: Patent
Japanese
  FAMILY ACC. NUM. COUNT:
PATENT INFORMATION:
                   PATENT NO.
                                                                                         KIND
                                                                                                               DATE
                                                                                                                                                         APPLICATION NO.
                                                                                                                                                                                                                                         DATE
                   JP 2002263505
                                                                                                                20020917
                                                                                                                                                         JP 2001-71781
                                                                                                                                                                                                                                        20010314
                                                                                                                                                         TW 2002-9110403
CA 2002-2440602
WO 2002-JP2102
                                                                                                                                                                    2002-91104030
                    TW 224523
CA 2440602
                                                                                                                  20041201
                                                                                            A1
A1
                                                                                                                 20020919
                                                                                                                                                                                                                                         20020307
                     WO 2002072255
                                                                                                                20020919
                                                                                                                                                                                                                                        20020307
                  MO 2002072255 A1 20020919 WO 2002-072102 20020307

N: AE, AG, AL, AM, AT, AU, AZ, BA, BB, BB, BB, BY, BZ, CA, CH, CN,
CO, CR, CU, CC, DE, DK, DM, DZ, EC, EE, ES, FI, GB, GD, GE, GH,
GM, HR, HU, ID, IL, IN, IS, KE, KG, KP, KR, KZ, LC, LK, LR, LS,
LT, LU, LV, MA, MD, MG, MK, MN, MW, KK, MZ, NO, MZ, OM, PH, PL,
PT, RO, RU, SD, SE, SG, SI, SK, SL, TJ, TM, TN, TR, TT, TZ, UA,
UG, US, UZ, VM, YU, ZA, ZM, ZW

RN: GH, GM, KE, LS, MN, MZ, SD, SL, SZ, TZ, UG, ZM, ZW, AT, BE, CH,
CY, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE, TR,
BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG
AU 2002236240 A1 20020924 A1 20020307

R: AT, BE, CH, DS, DK, ES, FR, GB, GR, IT, LI, U, N, SE, MC, PT,
R: AT, BE, CH, DS, DK, ES, FR, GB, GR, IT, LI, U, N, SE, MC, PT,
                  AU 2002236240
EP 1371414
A1 20031217
EP 2002-702781
20020307
R: AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT,
IE, SI, LT, LV, FI, RO, MK, CY, AL, TR
BR 2002008058
A 20040302
BR 2002-80581
A 20040302
BR 2002-805614
20020307
2003-9014211
20030911
                    CN 1501839
US 2004082800
                                                                                                                 20040429
                                                                                                                                                         US 2003-471421
                                                                                                                                                                                                                                        20030911
                     US 6982235
                                                                                                                20060103
20051125
  IN 2003CN01449
PRIORITY APPLN. INFO.:
                                                                                                                                                         IN 2003-CN1449
JP 2001-71781
                                                                                                                                                                                                                             20030915
A 20010314
                                                                                                                                                         WO 2002-JP2102
                                                                                                                                                                                                                              w 20020307
```

```
ANSWER 3 OF 8 CAPLUS COPYRIGHT 2007 ACS on STN

In a system for manufacturing propylene oxide by spoxidn. of
liquid propylene (I) with liquid organic hydroperoxide

in the presence of a catalyst, ≥2 pumps are equipped in parallel in
a passage, through which I is supplied. In this system, supply of I is
ensured, thus preventing deactivation of the catalyst even in an
eency
emergency
case where one of the I-supplying pumps is terminated.
ACCESSION NUMBER: 2003:274775 CAPLUS
DOCUMENT NUMBER: 138:272089
TITLE: System for manufacturing propylene
                                                                      138:272089
System for manufacturing propylene oxide and its manufacture
Katao, Masaaki: Omae, Shunichi; Shinohara, Koji Sumitomo Chemical Co., Ltd., Japan Jpn. Kokai Tokkyo Koho, 3 pp.
CODEN: JKXXAF
Patent
Japanese
 INVENTOR(S):
 PATENT ASSIGNEE(S):
SOURCE: .
DOCUMENT TYPE:
LANGUAGE:
FAMILY ACC. NUM. COUNT:
PATENT INFORMATION:
                                                                                                                                                                                               DATE
               PATENT NO.
                                                                        KIND
                                                                                      DATE
                                                                                                                              APPLICATION NO.
JP 2003104979
PRIORITY APPLN. INFO.:
                                                                         A
                                                                                           20030409
                                                                                                                              JP 2001-299008
JP 2001-299008
                                                                                                                                                                                                20010928
```

```
ANSMER S OF 8 CAPLUS COPYRIGHT 2007 ACS on STN
Titanovanadosilicalites are very selective and active catalysts in the
epoxidn. of olefins by peroxides. Diluted H202 suffices to afford
high yields of the epoxide. V incorporation at levels of $i:V = 100-2500
effectively changes the characteristics of the titanosilicalite into
effectively changes the unside terms of the which it is incorporated to give near quant, conversion of propylene at selectivities >90%. For example, reacting liquid propylene with H202 (30% aqueous solution) in MeOM for 6 h at 35*7500 pai under N in the presence of K-exchanged Ti-V-silicalite catalyst (average particle size 130 nm; preparation given) gave 95% propylene oxide with propylene conversion >9%.

ACCESSION NUMBER: 1998:263255 CAPLUS
DOCUMENT NUMBER: 128:321554

Titanovanadosilicalites as epoxidation
 DOCUMENT NUMBER:
TITLE:
                                                      128:321554
Titanovanadosilicalites as epoxidation
catalysts for olefins
Nemeth, Laszlo T.: Lewis, Gregory J.: Rosin, Richard
 INVENTOR (S):
                                                     R.
UOP LLC, USA
U.S., 7 pp.
CODEN: USXXAM
Patent
 PATENT ASSIGNÉE(S):
SOURCE:
 DOCUMENT TYPE:
                                                      English
 LANGUAGE:
FAMILY ACC. NUM. COUNT:
PATENT INFORMATION:
            PATENT NO.
                                                      KIND
                                                                 DATE
                                                                                             APPLICATION NO.
                                                                                                                                               DATE
                                                                                             US 1997-818265
            US 5744619
                                                                    19980428
                                                                                                                                               19970317
                                                       A
            ZA 9806223
                                                                     19990202
                                                                                              ZA 1998-6223
            CA 2243009
CA 2243009
EP 978315
EP 978315
                                                                     20000113
                                                                                              CA 1998-2243009
                                                                     20070619
                                                                     20000205
                                                                                             EP 1998-305563
                                                                                                                                               19980713
                                                                     20030924
                   R: AT, BE, CH, DE,
IE, SI, LT, LV,
2206845 T3
                                                             DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT, FI, RO
            ES 2206845
                                                                     20040516
                                                                                             ES 1998-305563
                                                                                                                                               19980713
            IN 1998DE01993
                                                                    20060113
                                                                                              IN 1998-DE1993
CN 1998-103371
                                                                                                                                               19980713
            CN 1241564
                                                                     20000119
                                                                                                                                               19980714
            AU 9876141
                                                                     20000203
                                                                                              AU 1998-76141
US 1997-818265
                                                                                                                                               19980714
 PRIORITY APPLN. INFO.:
                                                                                                                                        A 19970317
                                                                                             US 1997-840531 '
                                                                                                                                       A 19970422
                                                                                             EP 1998-305563
                                                                                                                                       A 19980713
                                                                                                                                        A 19980714
```

OTHER SOURCE(S): CASREACT 128:321554
REFERENCE COUNT: 11 THERE ARE 11 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE

FORMAT

```
L1 ANSWER 6 OF 8 CAPLUS COPYRIGHT 2007 ACS on STN

AB Epoxides are prepared in the liquid phase by reacting an ethylenically unsatd.

compound with 1 part organic hydroperoxide in 4-20 parts anhydrous organic solvent at 80-160° in the presence of molybdate catalyst. The molybdate, which has good solubility in the organic medium, a high concentration in Mo, very high catalytic activity, weak acidity, and high purity, is present in a concentration of 10-4 to 2 * 10-3 mole/kg, solvent and hydroperoxide. Thus, 400 g. com. Mo03.820 containing 90% Mo03 was dissolved in 900 g. concentrated MC1 (d. 1.19) preheated to 90°, the mixture cooled to room temperature, the molybdic chloride separated from the reaction mixture by extracting twice with a total of 2 1. Et20, the ether solution dried and evaporated to give 905 g. colorless crystals, the crystals redissolved in dry ether, 440 g. propylene oxide in 500 cc. Et20 added to the solution at 10-15° during 3 hrs., the mixture stirred 1 hr. and the precipitate filtered off and washed with dry ether, water-saturated ether, and then dry ether and dried at 40° under vacuum to give 465 g. propylene glycol molybdate (Mo04C3H6) (f) containing 71.9% Mo03. I (f g.) was dissolved in 1 g. propylene plycol at 100°, the product mixed with 500 g. tert-BuOH, 500 g. 99% tert-BuOOH added to give a solution containing 5 * 10-3 g. atoms Mo/kg., 10 cc. of this solution and 20° cc. liquid propylene at -80° were sealed in a pressure-resistant glass tube, heated to 110°, cooled to -80°, and degassed to give a solution containing apprx.10% propylene exide with a 79% conversion of hydroperoxide.

ACCESSION NUMBER: 71:1417
ITTLE: Epoxides: molybdate catalysis PODCUMENT NUMBER: 71:1417
ITTLE: Epoxides: molybdate catalysis PODCUMENT TYPE: Patent Prench PAMILY ACC. NUM. COUNT: 1
PATENT NO. KIND DATE APPLICATION NO. DATE

DOCUMENT TYPE: Patent Prench PAMILY ACC. NUM. COUNT: 1
PATENT NO. KIND DATE APPLICATION NO. DATE
```

```
ANSWER 8 OF 8 CAPLUS COPYRIGHT 2007 ACS on STN
The title compds, are prepared by contacting C2-4 olefins with a C4-8
tert-alkyl hydroperoxide at 50-200° in an organic solvent
containing at least 20% by weight hydrocarbon in the presence of
allic Mo or a
Mo compound Thus, expts. Were carried out with 25 g. 94% tert-BuOOH and
0.05 g. Mo(CO)6 as catalyst while tert-BuOH and C6H6 were used as
                                 nt.
To this mixture was added 100 cc. liquid propylene and
the reaction carried out 1 hr. at 110-11. The following results
were obtained (tert-BuOH in g., C6H6 in g., C6H6 % by weight, conversion
                                   mole %, and yield of 1,2-epoxypropane in mole % given): 0, 125, 100,
   92.2,

88.8 (at a reaction temperature of 106*): 25, 100, 80, 82.0, 89.3: 50,
75, 60, 70.8, 84.8: 75, 50, 40, 58.3, 86.0: 100, 25, 20, 47.0, 86.5: 125
0, 0, 43.0, 77.2. A similar experiment with 25 g. tert-BuoOH, 0.05 g.

No(CO)6,
and 125 g. tert-BuOH and no hydrocarbon solvent gave, when treated with
100 cc. liquid propylene 1 hr. at 106*, 43.5
mole % conversion and 64.3 mole % yield of 1.2-epoxypropane. Under
optimum conditions a yield of 75 mole % and a conversion of 89 mole %
were
                                 obtained. Similarly, 22.4 g. tert-BuoOH (100 %), 22.4 g. tert-BuOH, 0.1 g. Mo(CO)6, 100 cc. liquid propylene allowed to react 1 hr. at 110-11' gave with 120 g. xylene (isomeric mixture) 93.7 mole % conversion and 70.0 mole % yield. The use of 140 g. xylene gave 91.7 mole % conversion and 80.2 mole % yield. The latter experiment carried
                                   with other catalysts gave the following results (amount of catalyst catalyst, conversion, and yield in mole \(\bar{a}\) given! 5.053 g., MoCl5, 9 20.201.5 g., MoCl5, 9 20.201.5 g., MoCl5, 9 20.201.5 g., MoCl (freshly prepared by reduction of NaZMoO4 with
## 2.07 1.5 g., MoO2 (freshly prepared ,
## 1.07 1.5 g., MoO2 (freshly prepared ,
## 2.07 1.5 g., MoO2 (freshly prepare
      FAMILY ACC. NUM. COUNT:
PATENT INFORMATION:
                                   PATENT NO.
                                                                                                                                                                KIND
                                                                                                                                                                                                         DATE
                                                                                                                                                                                                                                                                                   APPLICATION NO.
                                                                                                                                                                                                                                                                                                                                                                                                                                  DATE
                                   NL 6605820
DE 1568002
FR 89937
                                                                                                                                                                                                                                                                                                                                                                                                                                  19660429
                                                                                                                                                                                                         19670102
                                                                                                                                                                                                                                                                                   NL 1966-5820
      GB 1149344
PRIORITY APPLN, INFO.:
                                                                                                                                                                                                                                                                                                                                                                                                                                  19650701
```

```
ANSWER 7 OF 8 CAPLUS COPYRIGHT 2007 ACS ON STN
Olefins are contacted in the liquid phase with tert-BuoOH at
50-200° in the presence of a Mo metal catalyst whereby the ratio Mo
metal surface to the number of g. hydroperoxide is 1-20 cm.2/g.
Thus, 100 g. liquid propylene was contacted with 12.4
g. tert-BuoOH, 22.4 g. tert-BuOH, 140 g. xylene, and the Mo metal
catalyst. The following results were obtained (ratio cm.2/g., reaction
time, min. temperature, conversion in mel. 1, and yield of epoxide with
respect
to converted hydroperoxide given): 23.3, 60, 110-11*,
90.8, 64.7; 23.3, 20, 110-11*, 52.7, 72.5; 3.9, 60, 110-11*,
82.5, 75.2; 3.9, 20, 110-11*, 32.1, 90.5; 23.3, 60, 100-11*,
73.5, 74.7; 3.9, 60, 105-6*, 75.7, 79.2. A mixture containing 1.73 g.
1-octene, 0.513 g. tert-BuoOH, and a Mo metal plate with a total surface
of 1.8 cm.2 was heated at 102* and kept 20 min. at 102*
(ratio Mo metal to tert-BuoOH was 3.5 cm.2/g.) to give a conversion of 37
mole 4 and a yield of 100 mole 4.
ACCESSION NUMBER: 1967:432577 CAPLUS
DOCUMENT NUMBER: 61:32577
  DOCUMENT NUMBER:
ORIGINAL REFERENCE NO.:
                                                                                            67:32577
                                                                                          67:32577
67:6155a
Epoxides
Atlantic Refining Co.
Neth. Appl., 8 pp. Addn. to Neth. Appl. 6517166
CODEN: NAXXAN
  PATENT ASSIGNEE(S):
  SOURCE:
DOCUMENT TYPE:
  LANGUAGE
  LANGUAGE:
FAMILY ACC. NUM. COUNT:
PATENT INFORMATION:
                                                                                                                                                                APPLICATION NO.
                   PATENT NO.
                                                                                            KIND
                                                                                                                     DATE
                                                                                                                                                                                                                                                    DATE
NL 6605821
DE 1568001
FR 89938
GB 1146202
PRIORITY APPLN. INFO.:
                                                                                                                                                               NL 1966-5821
DE
FR
GB
US
                                                                                                                     19670102
                                                                                                                                                                                                                                                    19660429
                                                                                                                                                                                                                                                    19650701
```

=> FIL STNGUIDE

COST IN U.S. DOLLARS SINCE FILE TOTAL ENTRY SESSION

FULL ESTIMATED COST 69.29 69.50

DISCOUNT AMOUNTS (FOR QUALIFYING ACCOUNTS) SINCE FILE TOTAL

CA SUBSCRIBER PRICE ENTRY SESSION -6.24 -6.24

FILE 'STNGUIDE' ENTERED AT 17:38:29 ON 20 SEP 2007 USE IS SUBJECT TO THE TERMS OF YOUR CUSTOMER AGREEMENT COPYRIGHT (C) 2007 AMERICAN CHEMICAL SOCIETY (ACS)

FILE CONTAINS CURRENT INFORMATION.
LAST RELOADED: Sep 14, 2007 (20070914/UP).

=> log hold

COST IN U.S. DOLLARS . SINCE FILE TOTAL

FULL ESTIMATED COST ENTRY SESSION 0.48 69.98

DISCOUNT AMOUNTS (FOR QUALIFYING ACCOUNTS) SINCE FILE TOTAL

CA SUBSCRIBER PRICE ENTRY SESSION 0.00 -6.24

SESSION WILL BE HELD FOR 120 MINUTES
STN INTERNATIONAL SESSION SUSPENDED AT 17:43:11 ON 20 SEP 2007